

3-Acetoxyspathulenol, a New Aromadendrane-Type Natural Product from the Soft Coral *Parerythropodium fulvum*

Matthias Wessels, Gabriele M. König,[†] and Anthony D. Wright^{*†}

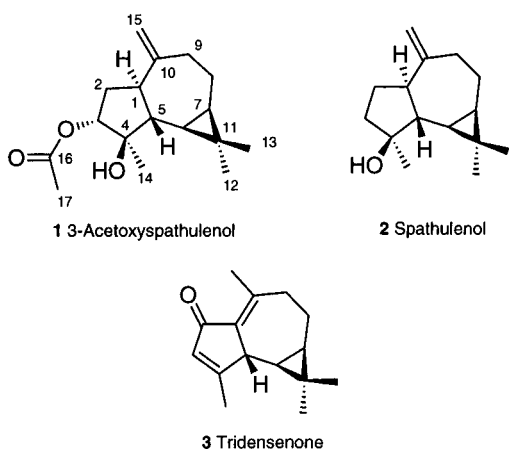
Institute for Pharmaceutical Biology, Technical University of Braunschweig, Mendelssohnstrasse 1, D-38106 Braunschweig, Germany

Received August 7, 2000

From the lipophilic extract of the soft coral *Parerythropodium fulvum* a new sesquiterpene with an aromadendrane-type carbon skeleton, 3-acetoxyspathulenol (**1**), was isolated, and the known compounds spathulenol (**2**) and tridensenone (**3**) were identified. The structure of the new compound was determined by interpretation of its spectroscopic data, including 1D and 2D ¹H and ¹³C NMR (COSY, HMQC, HMBC) and MS.

Chemical investigations have revealed soft corals as a rich source of secondary metabolites, mainly sesqui- and diterpenes.^{1,2} Some of these investigations focused on the ecological function of soft coral terpenoids and showed them to have important roles in interactions between marine organisms³ or indicated the terpenoid content to be useful in the taxonomic identification of soft corals.⁴ Other studies reported significant biological activities for several compounds.^{2,5} Terpenes of octocorallian origin have been shown to have antiinflammatory, antibacterial, and neuromuscular activities.² Many of the compounds are isolated from soft corals of the family Alcyoniidae, mainly from the genera *Sinularia*, *Sarcophyton*, and *Lobophytum*, with very few investigations being carried out on *Parerythropodium* sp.^{1,2}

In the present study the secondary metabolite content of the soft coral *Parerythropodium fulvum* (Forsk., 1775) (family Alcyoniidae), collected from the reef fringing Phantom Island, Great Barrier Reef, Australia, was investigated. The CH₂Cl₂ extract of this animal was found to contain the new compound, **1**, and the known compounds spathulenol (**2**) and tridensenone (**3**). Fractionation and purification of the CH₂Cl₂ extract were guided by TLC and ¹H NMR and were carried out by both vacuum liquid chromatography (VLC) and HPLC.



The ¹³C NMR and mass spectral data of **1** indicated it to have the molecular formula C₁₇H₂₆O₃ and thus have five elements of unsaturation. Analysis of its ¹³C NMR spectrum showed the presence of one carbonyl group, as an acetate (171.0 ppm), and one 1,1-substituted C–C double bond (152.8, s; 106.2, t, ppm), as the only multiple bonds within the molecule; thus, the molecule is tricyclic. Further analysis of the ¹³C NMR data revealed the presence of two further quaternary carbons (81.4 [bearing OAc or OH], 20.4 ppm), five methines (82.4 [bearing OAc or OH], 52.5, 48.3, 28.8, 27.5 ppm), three further methylenes (38.9, 31.4, 24.9 ppm), and four methyl groups (28.7, 21.2, 19.9, 16.2 ppm). Analysis of the ¹H NMR spectrum showed all the resonances for the methyl groups to be present as singlets (2.08, 1.22, 1.07, 1.03 ppm), indicating them to be bonded to quaternary carbons. Also evident from the ¹H NMR data was the presence of a cyclopropyl moiety (0.74 [ddd, 1H, *J* = 5.9, 9.4, 11.2 Hz], 0.58 [dd, 1H, *J* = 11.1, 9.4 Hz] ppm).

After all protons had been assigned to their directly bonded carbon atoms, aided by an HMQC spectrum, it was possible to deduce one molecular fragment. Thus, from the ¹H–¹H COSY spectrum, couplings in the form of cross-peaks were observed between H-3 and H₂-2, H₂-2, and H-1, and H-1 and H-5. Furthermore, H-5 coupled with H-6, and H-6 with H-7; H-7 also coupled with H₂-8, which further coupled with H₂-9, thus completing the molecular fragment H-3 to H₂-9, via H-5. This information in conjunction with the HMBC spectral data finally led to the planar structure of **1** being deduced. Thus, H₃-14 showed heteronuclear multiple bond couplings (HMBCs) with C-3, C-4, and C-5, indicating it to bond directly to C-4, which bonds to both C-3 and C-5, and in so doing completed the five-membered ring. Further, the ¹H NMR resonances of methyl groups CH₃-13 and CH₃-12 showed HMBC cross-peaks to the resonances of each other's C-atoms as well as to those of C-6, C-7, and C-11, thus completing the cyclopropyl ring. HMBC correlations observed from the resonances associated with H₂-15 to those of C-1 and C-9 positioned the exomethylene group between these two carbon atoms and in so doing gave rise to the third and final ring. Finally, the cross-peak between the resonance of H-3 and the resonance of C-16 clearly showed the acetate function to reside at C-3 and the OH group at C-4 (81.4 ppm).

To determine the relative stereostructure of **1**, NOE difference experiments were performed. Thus, irradiation at the resonance frequency of H-3 resulted in an enhancement of the resonances associated with H-5 and H₂-2, indicating H-3 and H-5 to be on the same side of the molecule. The magnitude of the coupling constants of H-5

* To whom correspondence should be addressed. E-mail: a.wright@uni-bonn.de.

[†] Current address: Institute for Pharmaceutical Biology, University of Bonn, Nussallee 6, 53115 Bonn, Germany; <http://www.tu-bs.de/institute/pharm.biol/GAWK.html>.

(δ 1.40 [dd, 1H, J = 9.8, 11.1 Hz]) indicated it to have two diaxial interactions and showed H-1 and H-6 to be on the same side of the molecule and on the opposite side from H-5. Irradiation at the resonance frequency of H-1 (δ 2.50 [ddd, 1H, J = 8.0, 9.6, 9.8 Hz]) resulted in NOEs being detected between it and H-6 and H₃-12 and showed them to be on the same side of the molecule and on the opposite side of H-3 and H-5. Further NOEs observed between H-6 and H-7, H-7 and H₃-12, H-6 and H₃-12, and H-5 and H₃-13 enabled all of the remaining stereochemistries to be assigned and the relative configuration of **1** to be deduced as shown in **1**.

Together with **1** (3-acetoxyspathulenol) the known compounds spathulenol (**2**) and tridensenone (**3**) were tentatively identified by GC-MS analysis of a fraction resulting from the isolation of **1**. The physical (GC-MS) and spectral (¹H NMR) data of **2** and **3** were in good agreement with those reported in the literature.⁶⁻⁸

Compound **1** is a new sesquiterpene with an aromadendrane-type carbon skeleton similar to the one reported from another investigation of *Parerythropodium* sp.⁹ In that study it was found that most of the sesquiterpenoid compounds were lost during freeze-drying or through simple evaporation, a phenomenon that has been reported earlier^{4,10} and which is directly related to the fact that soft corals have a high content of volatile compounds, mainly sesquiterpenes. Even though the currently investigated sample was freeze-dried, the three sesquiterpene compounds **1-3** were successfully isolated with apparent minimal, if any, losses as a result of evaporation probably due to the polar nature of their functionalities. Clearly, however, it is also possible that a great many other less polar metabolites were not found as a result of the sample preparation methodologies used. Compound **2**, first isolated from *Eucalyptus spathulata*,¹¹ has been described from a variety of other terrestrial plants too.¹² Compound **3** was also first isolated from a terrestrial source, the Japanese liverwort *Bazzania tridens*.⁸ This report is the first of spathulenol (**2**) and tridensenone (**3**) or stereoisomers thereof as being constituents of marine organisms.

Compounds **1-3** (**2** and **3** as part of a mixture) were tested for their antimicrobial activity against two bacteria (*Escherichia coli*, *Bacillus megaterium*), four fungi (*Eurotium repens*, *Fusarium oxysporum*, *Microbotryum violaceum*, *Mycotypha microspora*), and the green alga *Chlorella fusca*. Their abilities to inhibit the enzymes reverse transcriptase of the human immunodeficiency virus type 1 (HIV-1-RT) and p56^{lck} tyrosine kinase (TK) were also assessed. Finally, their effects toward a nematode (*Caenorhabditis elegans*) and brine shrimp (*Artemia salina*) were also investigated. In all of the applied test systems no positive activities were observed.

Since the current work was completed a paper by Liu et al.,¹³ which contained the structure of the new natural product *ent*-3 β -hydroxyspathulenol, was published. The NMR data of this compound further support the relative stereochemistry proposed for **1**.

Experimental Section

General Experimental Procedures. The general experimental procedures were carried out as previously described.¹⁴

Animal Material. *Parerythropodium fulvum* (Alcyoniidae) was collected by hand using scuba at depths between 2 and 8 m from the reef fringing Phantom Island, North Queensland, Australia, in September of 1990. Collected material was stored at -20 °C until used. A voucher specimen, number (NTM C13070), is stored at the Museum and Art Gallery of the Northern Territory, Darwin, Australia.

Table 1. ¹H NMR^a and ¹³C NMR^b Data of Compound **1**^c

position	δ ¹ H	δ ¹³ C
1	2.50 (ddd, 1H, J 8.0, 9.6, 9.8)	48.3 d ^d
2	2.27 (ddd, 1H, J 5.9, 9.6, 13.5)	31.4 t
	1.72 (ddd, 1H, J 4.7, 8.0, 13.5)	
3	4.95 (dd, 1H, J 4.7, 5.9)	82.4 d
4		81.4 s
5	1.40 (dd, 1H, J 9.8, 11.1)	52.5 d
6	0.58 (dd, 1H, J 9.4, 11.1)	28.8 d
7	0.74 (ddd, 1H, J 5.9, 9.4, 11.2)	27.5 d
8	2.00 (m, 1H)	24.9 t
	1.01 (m, 1H)	
9	2.44 (ddd, 1H, J 1.4, 6.4, 13.4)	38.9 t
	2.05 (m, 1H)	
10		152.8 s
11		20.3 s
12	1.07 (s, 3H)	28.7 q
13	1.03 (s, 3H)	16.2 q
14	1.22 (s, 3H)	19.9 q
15	4.68 (br s, 2H)	106.2 t
16		171.0 s
17	2.08 (s, 3H)	21.2 q

^a 300 MHz, δ in ppm relative to CHCl₃ in CDCl₃ = 7.26; J in Hz. ^b 75.5 MHz, δ in ppm relative to CDCl₃ = 77.0. ^c All assignments are based on extensive 1D and 2D NMR measurements (HMBC, HMQC, COSY). ^d Implied multiplicities determined by DEPT (s = C, d = CH, t = CH₂, q = CH₃).

Animal tissue was freeze-dried (dry wt 91.2 g) and exhaustively extracted with CH₂Cl₂ (1.2 L) and then with MeOH (1.2 L). Solvents were removed in vacuo and the extracts filtered through a pad (5 mm thick) of Si gel. The MeOH extract was extracted with CH₂Cl₂, and the two lipophilic fractions were combined to yield 8.9 g (9.8%) of CH₂Cl₂-soluble and 7.9 g (8.7%) of MeOH-soluble material. The CH₂Cl₂-soluble fraction was applied to a vacuum liquid column (VLC, Si gel) and gradient eluted from light petroleum ether to EtOAc and thereafter with MeOH to yield 12 fractions, each of 100 mL. Repeated column chromatography, including HPLC (Si gel, petroleum ether/EtOAc) of fraction 3, and combined fractions 5 and 6 afforded compounds **1-3**.

3-Acetoxyspathulenol (1): oil (14.6 mg, 0.016%); [α]_D -0.9° (c 0.29, CHCl₃); IR ν_{\max} 3470-3420, 2980, 2925, 2860, 1740, 1720, 1640, 1455 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) and ¹³C NMR (CDCl₃, 75.5 MHz), see Table 1; EIMS m/z 278 (10) [M⁺], 263 (5), 236 (4), 218 (24), 200 (53), 185 (26), 175 (32), 157 (27), 147 (22), 133 (20), 119 (18), 105 (24); HREIMS m/z 278.1875 (calcd for C₁₇H₂₆O₃, 278.1875).

Spathulenol (2): oil (\approx 1.3 mg, 0.0014%); with GC-MS and ¹H NMR in agreement with those published.^{6,7}

Tridensenone (3): oil (\approx 2.5 mg, 0.0027%); with GC-MS and ¹H NMR in agreement with those published.⁸

Biological Tests. The antimicrobial,¹⁵ RT inhibition,¹⁶ and TK inhibition¹⁷ assays were carried out as previously described, and the nematocidal and brine shrimp assays following the methods of Peters et al.¹⁸

Acknowledgment. Soft coral taxonomy was performed by Dr. P. Alderslade, Museum and Art Gallery of the Northern Territory, Darwin, Australia. We thank Mr. C. Dreikorn, TU-BS, for performing antimicrobial, TK, and the HIV-1-RT assays, and Dr. V. Wray and his group, GBF-Braunschweig, Mascheroder Weg 1, 38124 Braunschweig, for recording all NMR spectra. Thanks also go to Dr. U. Papke and Ms. D. Döring, MS-Service, Department of Chemistry, TU-BS, for making all MS measurements, and to Dr. L. Witte, TU-BS, for performing GC-MS analyses. Funding from the Bundesministerium für Bildung und Forschung (BMBF), Bayer AG, Leverkusen, and Fond der Chemischen Industrie is gratefully acknowledged.

References and Notes

- Faulkner, D. J. *Nat. Prod. Rep.* **1999**, *16*, 155-198, and references therein.

- (2) Krebs, H. Chr. *Prog. Chem. Org. Nat. Prod.* **1986**, *49*, 151–363.
- (3) Coll, J. C.; Bowden, B. F.; Heaton, A.; Scheuer, P. J.; Li, M. K. W.; Clardy, J.; Schulte, G. K.; Finer-Moore, J. *J. Chem. Ecol.* **1989**, *15*, 1177–1191.
- (4) Kashman, Y.; Loya, Y.; Bodner, M.; Groweiss, A.; Benayahu, Y.; Naveh, N. *Mar. Biol.* **1980**, *55*, 255–259.
- (5) Rodriguez, A. D.; Piña, I. C.; Barnes, C. L. *J. Org. Chem.* **1995**, *60*, 8096–8100.
- (6) Inagaki, F.; Abe, A. *J. Chem. Soc., Perkin Trans. 2* **1985**, 1773–1778.
- (7) Maurer, B.; Hauser, A. *Helv. Chim. Acta* **1983**, *66*, 2223–2235.
- (8) Toyota, M.; Asakawa, Y.; Takemoto, T. *Phytochemistry* **1981**, *20*, 2359–2366.
- (9) Green, D.; Kashman, Y.; Benayahu, Y. *J. Nat. Prod.* **1992**, *55*, 1186–1196.
- (10) Coll, J. C.; Bowden, B. F.; Clayton, M. N. *Chem. Br.* **1990**, 761–763.
- (11) Bowyer, J. C.; Jeffries, P. R. *Chem. Ind. (London)* **1963**, 1245–1246.
- (12) According to the Beilstein database (1999), there are 79 species in 57 genera described as sources of spathulenol (2). The genera belong mainly to the mosses, Asteraceae, and Lamiaceae.
- (13) Liu, H.-J.; Wu, C.-L.; Becker, H.; Zapp, J. *Phytochemistry* **2000**, *53*, 845–849.
- (14) Wright, A. D.; König, G. M.; Angerhofer, C. K.; Greenidge, P.; Linden, A.; Desqueyroux-Faundez, R. J. *J. Nat. Prod.* **1996**, *59*, 710–716.
- (15) Schulz, B.; Sucker, J.; Aust, H. J.; Krohn, K.; Ludewig, K.; Jones, P. G.; Döring, D. *Mycol. Res.* **1995**, *99*, 1007–1015.
- (16) Eberle, J.; Seibl, R. *J. Virol. Methods* **1992**, *40*, 347–356.
- (17) Wessels, M.; König, G. M.; Wright, A. D. *J. Nat. Prod.* **1999**, *62*, 927–930.
- (18) Peters, L.; König, G. M.; Wright, A. D. Unpublished data.

NP0003746